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On the Mark–Houwink parameters for galactomannans

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Abstract

We have produced a summary plot for a number of published results for the Mark–Houwink parameters for guar and other galactomannans. Agreement between these results, collected over almost 25 years, is very good, and suggests any of the reported parameter sets discussed would be equally valid to other workers. This also implies that any values measured outside the range described should be examined very critically.

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The so-called Mark–Houwink–Sakurada equation (1), a label often truncated to just the first two of these authors, but sometimes with the name of Staudinger appended, is one of the most fundamental in the characterisation of high molecular weight polymers, relating as it does the intrinsic viscosity of a polymer, $[\eta]$, to its (usually weight average) molecular weight (molar mass), $M_{\rm w}$

$$[\eta] = K' M_{\mathbf{w}}^{\alpha} \tag{1}$$

Each of the two parameters, K' and α reflect a combination of contributions, but K' is related essentially to the intrinsic or local chain flexibility, including the orientation of the bonds to and from the constituent monosaccharides, whereas the exponent α can reflect the chain geometry, branched, sphere, rod or coil and the solvent "quality" – poor, θ or Flory, or good solvent respectively. The various values of the exponent, for example 0.5 for a linear coil like macromolecule in a θ -solvent are well known and available in any introductory textbook of polymer science (Young & Lovell, 1991).

Although values of these parameters are listed in the standard Polymer Handbook (Brandrup, Immergut, &

Grulke, 2003) for a wide range of polymers and solvents, the data for carbohydrate polymers in aqueous solvents are less prevalent, and in some cases there are still disputed results. This can be seen, for example, by examining other lists of values for different carbohydrate polymers (e.g. Lapasin & Pricl, 1995).

One reason for this is that it is still by no means a trivial exercise to obtain "molecular solutions" of polysaccharides in water – some workers simply do not allow sufficient time for hydration, others use too mild conditions. As is well appreciated, hardly any "native" (as opposed to partially hydrolysed) polysaccharides dissolve readily in water – dextran may be one of these few – and both natural sample variations and the difference in hydration conditions increase the problems.

We have argued for some years (Richardson & Ross-Murphy, 1987; Robinson, Ross-Murphy, & Morris, 1982) based upon both macromolecular and semi-dilute solution rheological results, that guar gum, the most prevalent galactomannan in use, should be regarded as a model coil-like polymer. This is even though its intrinsic stiffness, as determined by chain persistence length, is significantly greater, by up to a factor 10×, than that for most synthetic coil polymers in common use (Patel, Picout, Ross-Murphy, & Harding, 2006).

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There are a considerable number of published values for the polysaccharide from guar, and a few for other galactomannans such as locust bean gum, tara and fenugreek, all of which differ in their galactose substitution pattern and degree. However, the point of this comment is to register the fact that sufficient values now exist that we can have some real confidence in the absolute values of K' and α .

Fig. 1 summarises this statement. It includes a number of published studies on guar (and as explained later for LBG and tara) and summarises them in the form of a Mark–Houwink plot.

The data have been collected over 25 years, and include the widely cited results of Robinson et al. (1982) a valuable series of measurements by Beer, Wood, and Weisz (1999) together with recent studies by a team including the current authors, for both guar (Picout, Ross-Murphy, Errington, & Harding, 2001) and tara/LBG (Picout, Ross-Murphy, Jumel, & Harding, 2002) and a careful study by Cheng, Brown, and Prud'homme (2002).

In most of these studies, $M_{\rm w}$ was measured directly by static light scattering, using either low angle or multi-angle apparatus, in some cases coupled to size exclusion chromatography (SEC) equipment. In the case of the Cheng study a neat recursive method was applied to size exclusion data without coupled scattering. Samples were prepared by a variety of methods, including the pressure-cell method (Patel et al., 2006; Picout et al., 2002) and by selective thermal and enzymatic degradation (Cheng et al., 2002). As can be seen the data cover a range of almost two decades in $M_{\rm w}$ and (the Beer data) go down close to the limit where coil-like behaviour begins to break down because there

Table 1
Mark-Houwink parameters from previously published work

α	$K'/\times 10^4/dL/g$	Source
0.72	5.13	Beer et al. (1999)
0.70	6.00	Picout et al. (2001)
0.74	3.72	Picout et al. (2002)
0.72	3.80	Robinson et al. (1982)
0.747	3.04	Cheng et al. (2002)

are less than say 10 statistical segments (as calculated from the persistence length).

Also included are the authors' own Mark-Houwink parameters summarised in Table 1.

There are several points to make here, and to help justify the purpose of this short comment.

1. The data are in very reasonable agreement, and no particular datum stands out for being well outside the expected range. That said, there are some definite trends within the data sets, the early but widely cited values from Robinson and co-workers lie slightly below the other sets, while those of Beer et al. lie slightly above. In fact these two sets have identical α values (slopes), but the Beer K' is circa. 35% greater – a large difference in linear terms but not when plotted as here in 3×2 decade logarithmic coordinates. The Cheng data are well positioned, as are both of the Picout lines. We could conclude that any of the K'/α pairs would do just as well. It also suggests there is little point repeating such experiments yet again, and values which lie outside these must be re-examined very critically. It would, of course, be

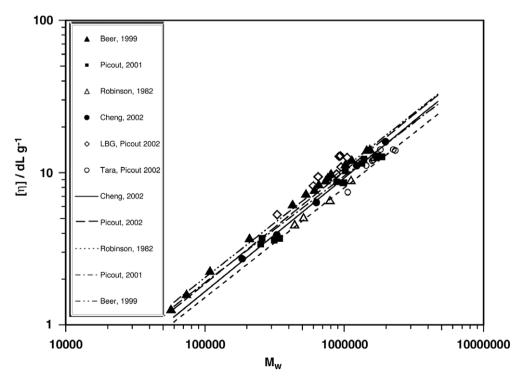


Fig. 1. Data for aqueous guar solutions (plus some for tara and LBG), from published work indicated; symbols as key.

- possible to calculate an overall "best" value for all the data, but we have chosen not to do so, since other authors may still have a personal preference.
- 2. Indeed, although there is very good agreement between these values, very different ones still appear in the literature, for example one paper (Wientjes, Duits, Jongschaap, & Mellema, 2000) reports $\alpha=1.05$ way outside the expected range for a linear polymer with dimensions perturbed by excluded volume (0.5–0.8). Although this, clearly mistaken result can be understood by re-examining the methodology in this paper (Lefebvre & Doublier, 2005; Picout et al., 2002), it was published in a quality journal, presumably without referees' comment.
- 3. The different Picout values correspond to guar samples (2001) and also are reanalysed including new data for tara and LBG (2002). As we concluded at the time (Picout et al., 2002) within experimental error there is no significant difference between the three different galactomannan samples. This suggests that the chain flexibility is sensitive only to the backbone geometry and is largely independent of the substitution amount and pattern. Surprising though this may be, it is actually consistent with our hypothesis based upon reanalysis (Picout & Ross-Murphy, 2002) of extensive data on cereal arabinoxylans, with the same β -1,4 backbone geometry (Dervilly-Pinel, Thibault, & Saulnier, 2001). It is also consistent with the classic measurements of the late Wyn Brown on cellulose in cadoxen (Brown & Wikström, 1965).
- 4. This leads us to a final point about the actual flexibility of such polysaccharides, in terms of persistent length. As we showed very recently (Patel et al., 2006) the absolute values appear to depend upon the precise method used for determination, but values for (all) the galactomannans lie in the range 4.1 ± 0.6 nm using the Burchard–Stockmayer–Fixman method (Picout et al., 2001), which assumes a contribution from excluded volume effects or 6.3 ± 0.2 nm, for guar and LBG alone, using the Bohdanecky method based on intrinsic chain stiffness (Bohdanecky, 1983) and the use of the Yamakawa–Fujii worm-like chain model. Very similar, if slightly smaller results, 3.1 ± 0.3 nm, were obtained by us (Picout & Ross-Murphy, 2002) for the arabinoxylan data (Dervilly-Pinel et al., 2001).

This also suggests some final thoughts on an equation closely related to Eq. (1) above, sometimes now called the Flory equation, which relates (formally the z-average root mean square) radius of gyration to $M_{\rm w}$.

$$\langle S^2 \rangle_z^{1/2} \equiv R_{\rm g} = K'' M_{\rm w}^{\nu} \tag{2}$$

As has been known for many years (Yamakawa, 1971), the exponent v for a linear high molecular chain in a θ -solvent (random flight chains) is 0.5, and in an excluded volume solvent it is ≈ 0.6 . A disappointing

number of early measurements using SEC-MALLS reported exponents v in the range say 0.1-0.3 for polysaccharides in aqueous/electrolyte solutions. More recent measurements, albeit by those "skilled in the art" (e.g. Mendichi, Doltes, & Schieroni, 2003; Vold, Kristiansen, & Christensen, 2006) have shown that realistic values of v can be obtained. The one *caveat* is that this is clearly not a trivial exercise, and all the raw data need to be examined seriously. It probably reflects the problems associated (sic) with obtaining molecular solutions of polysaccharides (Kratochvil, 1972). Here, it is well known that measurement of $[\eta]$ is less problematic.

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References

- Beer, M. U., Wood, P. J., & Weisz, J. (1999). A simple and rapid method for evaluation of Mark–Houwink–Sakurada constants of linear random coil polysaccharides using molecular weight and intrinsic viscosity determined by high performance size exclusion chromatography: application to guar galactomannan. *Carbohydrate Polymers*, 39, 377–380.
- Bohdanecky, M. (1983). New method for estimating the parameters of the wormlike chain model from the intrinsic-viscosity of stiff-chain polymers. *Macromolecules*, 16, 1483–1492.
- Brandrup, J., Immergut, E. H., & Grulke, E. A. (2003). *Polymer Handbook*. New York, NY: John Wiley and Sons Ltd.
- Brown, W., & Wikström, R. (1965). Viscosity-molecular weight relation for cellulose in cadoxen and a hydrodynamic interpretation. *European Polymer Journal*, 1, 1–10.
- Cheng, Y., Brown, K. M., & Prud'homme, R. K. (2002). Preparation and characterization of molecular weight fractions of guar galactomannans using acid and enzymatic hydrolysis. *International Journal of Biological Macromolecules*, 31, 29–35.
- Dervilly-Pinel, G., Thibault, J. F., & Saulnier, L. (2001). Experimental evidence for a semi-flexible conformation for arabinoxylans. *Carbo-hydrate Research*, 330, 365–372.
- Kratochvil, P. (1972). Particle scattering functions. In M. Huglin (Ed.), Light scattering from polymer solutions (pp. 333–384). London, UK: Academic Press.
- Lapasin, R., & Pricl, S. (1995). Rheology of industrial polysaccharides: Theory and applications. Glasgow, UK: Blackie Academic and Professional.
- Lefebvre, J., & Doublier, J. L. (2005). Rheological behaviour of polysaccharide aqueous systems. In S. Dimitriu (Ed.), *Polysaccharides:* Structural diversity and functional versatility (pp. 357–392). New York, NY: Marcel Dekker.
- Mendichi, R., Doltes, L., & Schieroni, A. G. (2003). Evaluation of radius of gyration and intrinsic viscosity molar mass dependence and stiffness of hyaluronan. *Biomacromolecules*, 4, 1805–1810.
- Patel, T. R., Picout, D. R., Ross-Murphy, S. B., & Harding, S. E. (2006). Pressure cell assisted solution characterisation of galactomannans 3. Application of analytical ultracentrifugation techniques. *Biomacro-molecules*, 7, 3513–3520.
- Picout, D. R., & Ross-Murphy, S. B. (2002). On the chain flexibility of arabinoxylans and other β-(1-4) polysaccharides. *Carbohydrate Research*, 337, 1781–1784.

- Picout, D. R., Ross-Murphy, S. B., Errington, N., & Harding, S. E. (2001). Pressure cell assisted solution characterisation of polysaccharides I: guar gum. *Biomacromolecules*, 2, 1301–1309.
- Picout, D. R., Ross-Murphy, S. B., Jumel, K., & Harding, S. E. (2002). Pressure cell assisted solution characterisation of polysaccharides 2: locust bean gum and tara gum. *Biomacromolecules*, 3, 761–767.
- Richardson, R. K., & Ross-Murphy, S. B. (1987). Nonlinear viscoelasticity of polysaccharide solutions. 1. Guar galactomannan solutions. *International Journal Biological Macromolecules*, 9, 250–256.
- Robinson, G., Ross-Murphy, S. B., & Morris, E. R. (1982). Viscosity molecular-weight relationships, intrinsic chain flexibility, and dynamic

- solution properties of guar galactomannan. Carbohydrate Research, 107, 17-32.
- Vold, I. M. N., Kristiansen, K. A., & Christensen, B. E. (2006). A study of the chain stiffness and extension of alginates, in vitro epimerized alginates, and periodate-oxidized alginates using size-exclusion chromatography combined with light scattering and viscosity detectors. *Biomacromolecules*, 7, 2136–2146.
- Wientjes, R. H. W., Duits, M. H. G., Jongschaap, R. J. J., & Mellema, J. (2000). Linear rheology of guar gum solutions. *Macromolecules*, 33, 9594–9605.
- Yamakawa, H. (1971). *Modern theory of polymer solutions*. New York, NY: Harper and Row.
- Young, R. J., & Lovell, P. A. (1991). *Introduction to polymers*. Kingston-on-Thames, UK: Nelson Thornes Ltd.